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Mr. Randy Matty, P.E. Air Management Engineer Wisconsin Department of Natural Resources 2984 Shawano Ave. Green Bay, WI 54313-6727 August 19, 2014

RE: B28 and B09 Compliance Stack Testing Notice

Dear Mr. Matty:

Manitowoc Public Utilities is planning to repeat the biennial PM stack testing of boilers B28 and B09 the week of September 8, 2014 as required by permit conditions LD.l.b.(4) and LE.l.b.(7). The Emission Compliance Test Protocol No. 4784B prepared by AIRTECH Environmental Services Inc. for Manitowoc Public Utilities previously submitted will be used for this test.

The proposed stack testing schedule is as follows:

- September 8<sup>th</sup> (Monday) Test team travel to and set up at MPU (Arrive by 12:00 pm)
- September 9<sup>th</sup> (Tuesday) B28 Compliance particulate matter stack testing, PM2.5, total PM
  Compliance PM Stack Test (3-runs) @ 8:00 am 5:00 pm
- September 10<sup>th</sup> (Wednesday) B9 Compliance particulate matter stack testing, PM2.5, total PM
  - o Compliance PM Stack Test (3-runs) @ 8:00 am − 5:00 pm
- September 11<sup>th</sup> (Thursday) Pack up and go @ 8:00 am, Make-up day if needed

Please call if you have any questions regarding the stack testing, or require additional information.

Sincerely,

Thomas E. Reed, P.E.

Environmental Engineer

Manitowoc Public Utilities

Phone: 920-686-4384

Cell: 920-973-7134

Fax: 920-686-4348

Cc: Red Jones – MPU

Jerry Ahlswede – MPU

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Tim Harding - MPU

Scott Karbon - MPU

Adam Becker – MPU

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File: PowerPlant/Regulatory - Environmental/Correspondence/WDNR Document: 2014-08-19 (WDNR) B8 and B9 Stack Test Notice

8/19/2014



# **Protocol for the Air Emissions Test Program**

To be conducted for Manitowoc Public Utilities At the Manitowoc Power Plant Facility ID 436035930 Located at 701 Columbus Street Manitowoc, Wisconsin 54220

> Protocol No. 4784B August 13, 2014

# **Table of Contents**

PROJECT OVERVIEW	
General	
Methodology	
Particulate Methodology	Error! Bookmark not defined.
Parameters	
TEST SCHEDULE	4
TEST PROCEDURES	5
Method Listing	5
Method Descriptions	5
Method 1	5
Method 2	5
Method 3	
Method 4	6
Method 19	7
	7
DESCRIPTION OF INSTALLATION	

### APPENDIX

Figures Sample Calculations Sample Calibration Sheets Sample Data Sheets

# **Project Overview**

### General

Airtech Environmental Services Inc. (Airtech) has been contracted by Manitowoc Public Utilities (MPU) to perform an air emissions test program at the MPU Power Plant (Facility ID 436035930) located in Manitowoc, Wisconsin. The specific objectives of this test program are as follows:

 Perform compliance testing to determine the concentrations of total filterable particulate matter (FPM), FPM less than a nominal aerodynamic diameter of 2.5 microns (PM<sub>2.5</sub>) and condensable particulate matter (CPM) from the exhausts of two (2), circulating, fluidized-bed boilers designated as Boiler 8 (B28) and Boiler 9 (B09)

Testing will be performed to meet the requirements of MPU; the Wisconsin Department of Natural Resources (WDNR); and the United States Environmental Protection Agency (US EPA), as applicable.

Testing will be conducted by Airtech Environmental Services Inc. Coordinating the test program will be:

Thomas Reed	Brandon Check
Manitowoc Public Utilities	Airtech Environmental Services Inc.
701 Columbus Street	1371 Brummel Ave.
Manitowoc, WI 54220	Elk Grove Village, IL 60007
Phone: (920) 686-4384	Phone: (630) 860-4740
email: tomreed@mpu.org	email: bcheck@airtechenv.com

### Methodology

EPA Method 5 combined with EPA Method 202 will be used to determine the concentration of FPM, filterable PM<sub>2.5</sub> and CPM at the exhaust of boilers B28 and B09. For this project, all FPM collected with the Method 5 fraction of each sampling train will be assumed to be filterable PM<sub>2.5</sub>. The sum of the FPM and CPM will be considered representative of the total particulate matter (PM) results.

In EPA Methods 5/202, a sample of the gas stream is withdrawn isokinetically from the source. FPM will collect in a glass probe and on a glass fiber filter. CPM will pass through the probe and filter assemblies and collect in a dry impinger system. Results for PM will be expressed in units of grains per dry standard cubic foot (gr/dscf), pounds per hour (lb/hr) and in units of pounds per million British thermal units (lb/MMBtu). Analysis for PM will be conducted at the Airtech laboratory located in Elk Grove Village, Illinois. Three (3), approximately one-hundred-twenty (120) minute test runs will be performed at each test location such that a minimum of 60 dscf is sampled.

In order to convert the various concentrations of PM to mass emissions rates, the volumetric gas flow rate through the test location will be determined concurrently with each test run, using EPA Methods 1, 2, 3B and 4.

### **Special Considerations**

The compliance PM testing of Boilers 28 and 09 will be conducted:

- 1. The configuration for the particulate train to determine total particulate emissions will be as follows:
  - Glass lined probe (heated to 320F +/- 25F)
  - Filter (heated to 320F +/- 25F)
  - Impinger train setup per USEPA Method 202 requirements
- 2. Sampling will be performed at the Method 1 required test points with a minimum of 60 dry standard cubic feet sampled, and an isokinetic variance of between 90%-110%.
- 3. Compliance testing to consist of three runs, each of which will be a minimum of 120 minutes.
- 4. Probe, train and sample recovery jars will be constructed of glass.
- 5. All laboratory recovery and analysis vessels will be constructed of glass.

6.

### **Parameters**

The following parameters will be determined at the exhaust of Boilers 28 and 09:

- oxigen concentration
- carbon dioxide concentration
- filterable particulate matter concentration
- condensable particulate matter concentration



# **Test Schedule**

Test dates to be determined.

Date	Location	Activity	<b>Test Method</b>	No. of Runs
1	Manitowoc, WI	Mobilize to job site		
	Boiler 9 (B09)	Set up test equipment		
2	Boiler 9 (B09)	Perform compliance PM testing	EPA M1, 2, 3B, 4 & 5/202	3 (120 min. each)
3	Boiler 8 (B28)	Set up test equipment		
4	Boiler 8 (B28)	Perform compliance PM testing	EPA M1, 2, 3B, 4 & 5/202	3 (120 min. each)
		Breakdown test equipment		
		Demobilize from job site		
5		Contingency Day		

### **Test Procedures**

### **Method Listing**

The test methods found in 40 CFR Part 60, Appendix A and 40 CFR Part 51, Appendix M will be referenced for the test program. The following individual methods will be referenced:

Method 1	Sample and velocity traverse for stationary sources
Method 2	Determination of stack gas velocity and volumetric flow rate (type S pitot tube)
Method 3	Determination of oxygen and carbon dioxide concentrations in emissions from stationary sources
Method 4	Determination of moisture content in stack gases
Method 5	Determination of particulate matter emissions from stationary sources
Method 19	Determination of sulfur dioxide, nitrogen oxide and carbon monoxide emission rates
Method 202	Dry impinger method for determining condensable particulate emissions from stationary sources

### **Method Descriptions**

### Method 1

EPA Method 1 will be used to determine the suitability of the sampling locations and to determine the traverse points used for the volumetric flow rate and PM determinations. The test locations must conform to the minimum requirement of being at least 2.0 diameters downstream and at least 0.5 diameters upstream from a flow disturbance.

### Method 2

EPA Method 2 will be used to determine the gas velocity through each test location using a type S pitot tube and an incline plane oil manometer. The manometer will be leveled and "zeroed" prior to each test run. The sample train will be leak checked before and after each run by pressurizing the positive side, or "high" side, of the pitot tube and creating a deflection on the manometer of at least three in.  $H_2O$ . The leak check will be considered valid if the manometer remains stable for 15 seconds. This procedure will be repeated on the negative side too. The velocity head pressure and gas temperature will then be determined at each point specified in Method 1. The static pressure of the duct will be measured using a water filled U-tube manometer. In addition, the barometric pressure will be measured and recorded. A diagram of the Method 2 apparatus is shown as part of the Method 5/202 sample train in Figure 3 of the Appendix.



### Method 3

The carbon dioxide and oxygen contents will be determined at the exhaust of B28 and B9 using EPA Method 3. A gas sample will be collected into a Tedlar bag from the back of each 5/202 sample train for the duration of each test run. Analysis will be performed using an Orsat gas analyzer within eight hours of collection. The Orsat and operator will be validated by analyzing at least one bag filled with an EPA protocol mixture of oxygen and carbon dioxide in a balance of nitrogen.

The analyzer will be leak checked prior to analysis by raising the liquid levels in each pipette to a reference mark on the capillary tubes and then closing the pipette valves. The burette solution will then be raised to bring the meniscus onto the graduated portion of the burette and the manifold valve will be closed. A leak check will be considered valid if the pipette meniscus does not fall below the reference mark and the burette meniscus does not fall by more than 0.2 ml after four minutes.

The carbon dioxide content and oxygen content will be used to calculate the dry molecular weight of the gas stream. The molecular weight will then be used, along with the moisture content determined by EPA Method 4, for the calculation of the volumetric flow rate. For these calculations, the balance of the gas stream is assumed to be nitrogen since the other gas stream components are insignificant for the purposes of calculating molecular weight.

### Method 4

The moisture content at each test location will be determined using EPA Method 4. A known volume of sample gas will be withdrawn from the source and the moisture will be condensed and measured. The dry standard volume of the sample gas will then be compared to the volume of moisture collected to determine the moisture content of the sample gas.

EPA Method 4 will be operated in conjunction with EPA Methods 5/202. To condense the water vapor, the gas sample will pass through a series of glass impingers. The impingers will be charged as outlined in method 5/202.

The sample train will be leak checked prior to the test run by capping the probe tip and pulling a vacuum greater than the highest vacuum expected during the test run. A leak check will be considered valid if the leak rate is less than 0.02 cubic feet per minute or four percent of the average sample rate. Sample gas will then be withdrawn from the source at a rate not exceeding 0.75 cubic feet per minute such that a minimum sample volume of 21 dry standard cubic feet will be collected. The gas meter reading, gas meter inlet and outlet temperatures, gas meter static pressure and pump vacuum will be recorded every five minutes during each test run.

After the test run the sample train will be leak checked at the highest vacuum encountered during the test run. The amount of water collected in the condenser system will be measured volumetrically with a graduated cylinder and the silica gel weight gain will be



determined gravimetrically. The net weight gain of water will be converted to a volume of wet gas and then compared to the amount of dry gas sampled to determine the moisture content. The moisture content will be used, along with the O<sub>2</sub> and CO<sub>2</sub> concentrations determined using EPA Method 3, for the calculation of the volumetric flow rate.

### Method 19

EPA Method 19 will be used to calculate pollutant emission rates in terms of pounds per million Btu (lb/MBtu). The calculation will be based on the oxygen content of the sample gas and an appropriate F factor, which is the ratio of combustion gas volumes to heat inputs.

### EPA Methods 5/202

EPA Methods 5 and 202 will be used to determine the particulate concentration at the exhaust of boilers B09 and B28. A sample of the gas stream will be withdrawn isokinetically from the stack and the particulate matter in the sample gas stream will collect in a glass probe and on a glass fiber filter. In addition, condensible particulate matter will be collected in a gas condenser system. The weight of particulate collected with the sample train combined with the volume of dry gas withdrawn from the stack is then used to calculate the total particulate concentration. Concentrations will be determined for the suspended particulate collected in the front-half of the sample train as well as the organic and inorganic condensable particulate collected in the back-half of the sample train. A diagram showing the major components of the sample train used for EPA Methods 2, 3, 4, 5 and 202 is shown in Figure 3 of the appendix.

To prevent contamination, all components of the sample train will be constructed of glass with no metal connections. Prior to testing the nozzle, probe and front-half of the filter holder will be washed using detergent and then rinsed with tap water, deionized water and lastly with acetone. The back-half of the filter holder, impingers and connecting glassware will be washed using detergent and then rinsed with tap water, deionized water and lastly with methylene chloride. After drying, all components will be sealed with parafilm or Teflon tape.

The sample probe to be used will consist of a glass liner and glass nozzle. Sample gas will pass through the nozzle and probe assembly and then through a glass fiber filter heated to 320°F (+/- 25°F). After exiting the filter, the sample gas will pass through the condenser system described in Method 4. The dry gas exiting the moisture condenser system will then pass through a sample pump and a dry gas meter to measure the gas volume. After leaving the dry gas meter the sample stream will pass through an orifice which will be used to meter the flow rate through the sample train. The pressure drop across the orifice will be measured with an incline plane oil manometer.

85mm Whatman glass fiber filters will be used as the substrate for the particulate sampling. The filter will be loaded into a glass filter holder with a Teflon support screen that was prepared in the same manner as the other components of the sample train. Prior to the test run, the filter will be desiccated for at least 24 hours and then weighed to the



nearest 0.0001g until a constant weight is achieved. The weight of the filter will be considered constant only when two consecutive weights taken at least six hours apart are within 0.0005g of each other. To eliminate interference in establishing a constant weight, both the analytical balance and the desiccators will be equipped with an ion generating polonium strip designed to eliminate static electricity that may collect on the samples.

The probe will be thoroughly pre-cleaned with acetone and the probe wash saved as a quality assurance check. The condenser system will then be prepared as outlined in Method 4. The sample train will be leak checked prior to the test run by capping the probe tip and pulling a vacuum of at least 15 in.Hg. A leak check will be considered valid if the leak rate is below four percent of the average sample rate or approximately 0.02 cfm. When not in operation inside the stack, the nozzle will be sealed with Teflon tape.

The probe tip will then be placed at each of the sample points determined in Method 1. The velocity at the sample point will be determined using Method 2 by reading the velocity pressure from the manometer. Sample will be withdrawn from the source at a rate such that the velocity at the opening of the nozzle is equivalent to the velocity of the stack gas at the sample point (isokinetically). During the test run the train will be moved to each of the Method 1 sample points. The sample time at each point will be calculated based on the number of sample points and the maximum allowable run time. The gas velocity pressure ( $\Delta P$ ), gas meter reading, gas meter inlet and outlet temperatures, gas meter orifice pressure ( $\Delta H$ ) and pump vacuum was recorded for each sample point. Each test run will at least one hour in duration.

After the test run the train will be leak checked at the highest vacuum encountered during the test run. The probe liner and nozzle will be rinsed with acetone and the rinse saved in a pre-cleaned amber glass jar with a Teflon lined lid. The condensate weight gain of the impinger contents will be determined as outlined in Method 4. The impingers and all connecting glassware will be rinsed twice with DUIF water. The condensate and rinse will be saved in a pre-cleaned amber glass jar with a Teflon lined lid. A final rinse of the above components will be performed with methylene chloride and saved in a separate a pre-cleaned amber glass jar with Teflon lined lid. The filter will be removed from the filter holder and placed in a labeled petri dish.

Analysis of the samples for particulate matter will be performed at Airtech located in Elk Grove Village, Illinois. The probe rinses will be transferred to a tared Teflon beaker liner, evaporated to dryness under ambient temperature and pressure conditions, desiccated for 24 hours and weighed to a constant weight. The filters will be desiccated for 24 hours and weighed to a constant weight. The weight gain of the probe rinse and filters yield the total weight of suspended particulate collected.

The collected condensate and rinse will be extracted by adding the contents of the methylene chloride rinse to the impinger water, and separating the layers in a separatory funnel. Two additional 75 ml portions of methylene chloride will be added to the funnel

to complete the extraction. The organic extract fraction will be placed into a tared beaker and evaporated at room temperature to dryness. The beakers will then be desiccated for 24 hours and weighed to a constant weight. The aqueous inorganic fraction will be placed into a tared beaker, taken to dryness at a slightly elevated temperature, and allowed to air dry at room temperature. The residue will be desiccated for 24 hours and weighed to a constant weight. The weight differentials for the organic and inorganic fractions will be combined to determine the total condensible particulate collected. All fractions of the particulate analysis will be adjusted for the appropriate blank values.



# **Description of Installation**

Manitowoc Public Utilities (MPU) is an electric cogenerating facility located in the City of Manitowoc Wisconsin. This plant includes two atmospheric pressure, circulating fluidized bed (CFB) boilers, designated as Boilers 8 (B28) and 9 (B09). Boiler 8 was installed in 1990, and is permitted to fire coal, petroleum coke, paper pellets, biomass, rubber waste derived fuels, natural gas, or other alternative fuels as approved by the Department. The Foster Wheeler Fluidized Bed Boiler is rated at 200,000 lbs. of superheated steam per hour at 975 psig and 905 degrees F. It is equipped with an economizer and air preheater and exhausts through a baghouse.

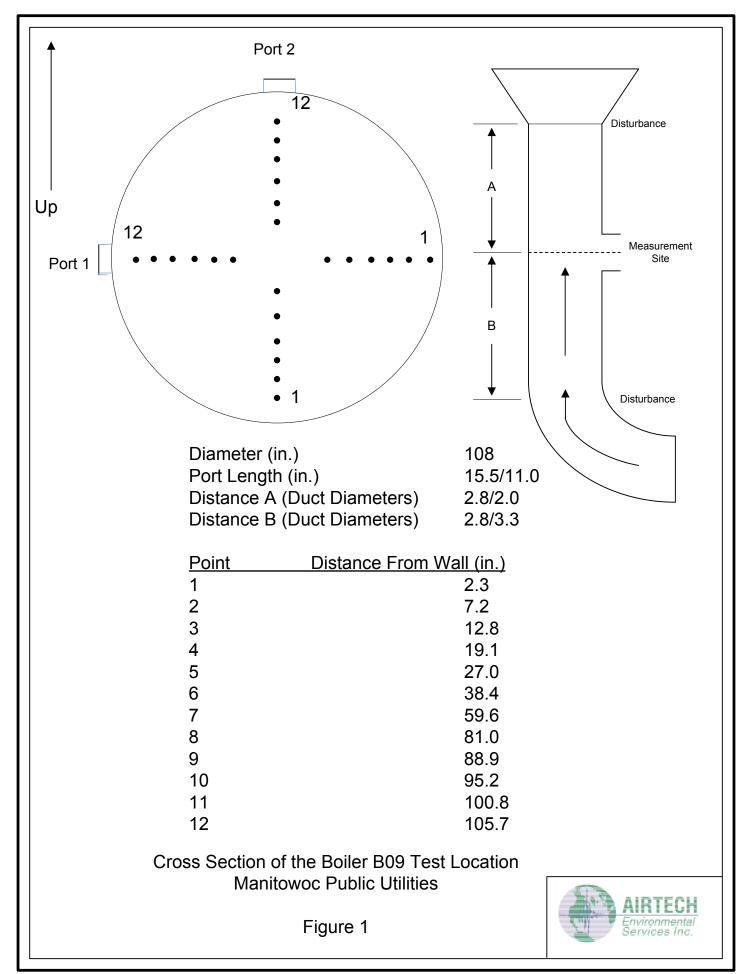
Boiler 8 (B28) was approved by the WDNR to be stack tested in the ductwork. A diagram for the unit ductwork is attached with the method 1 in the appendix. These diagrams show the breaching from the outlet of the I.D. fan to the stack outlet for B28. The test port detail drawing is also provided.

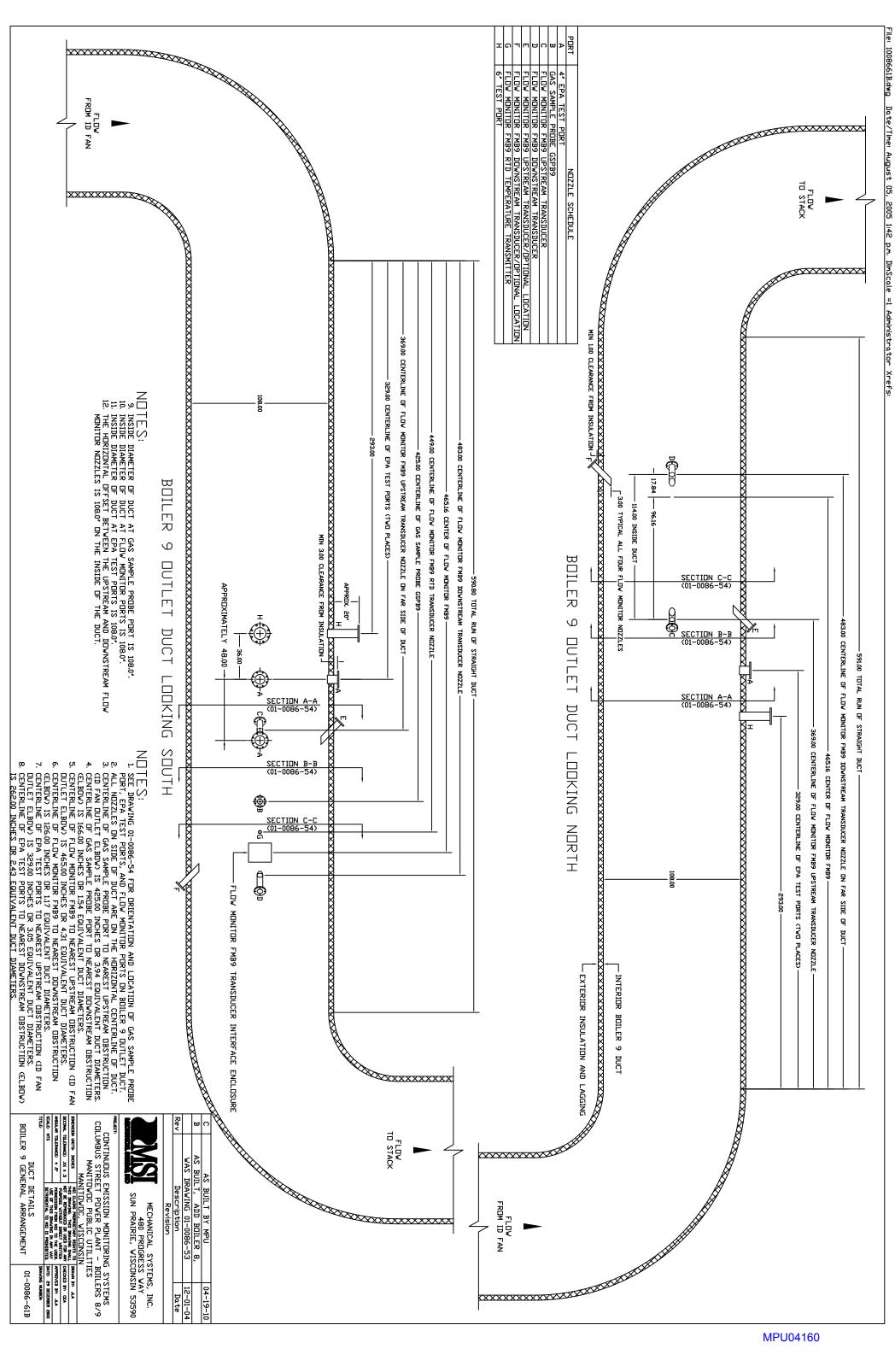
Boiler 9 (B09) was installed in 2004, and is permitted to fire coal, petroleum coke, renewable biomass and natural gas (start-up and load stabilization.) The Kvaerner/Mesto Fluidized Bed Boiler is rated at 475,000 lbs. of superheated steam per hour at 1,500 psig and 1,000 degrees F. It is equipped with an air preheater and exhausts through a baghouse.

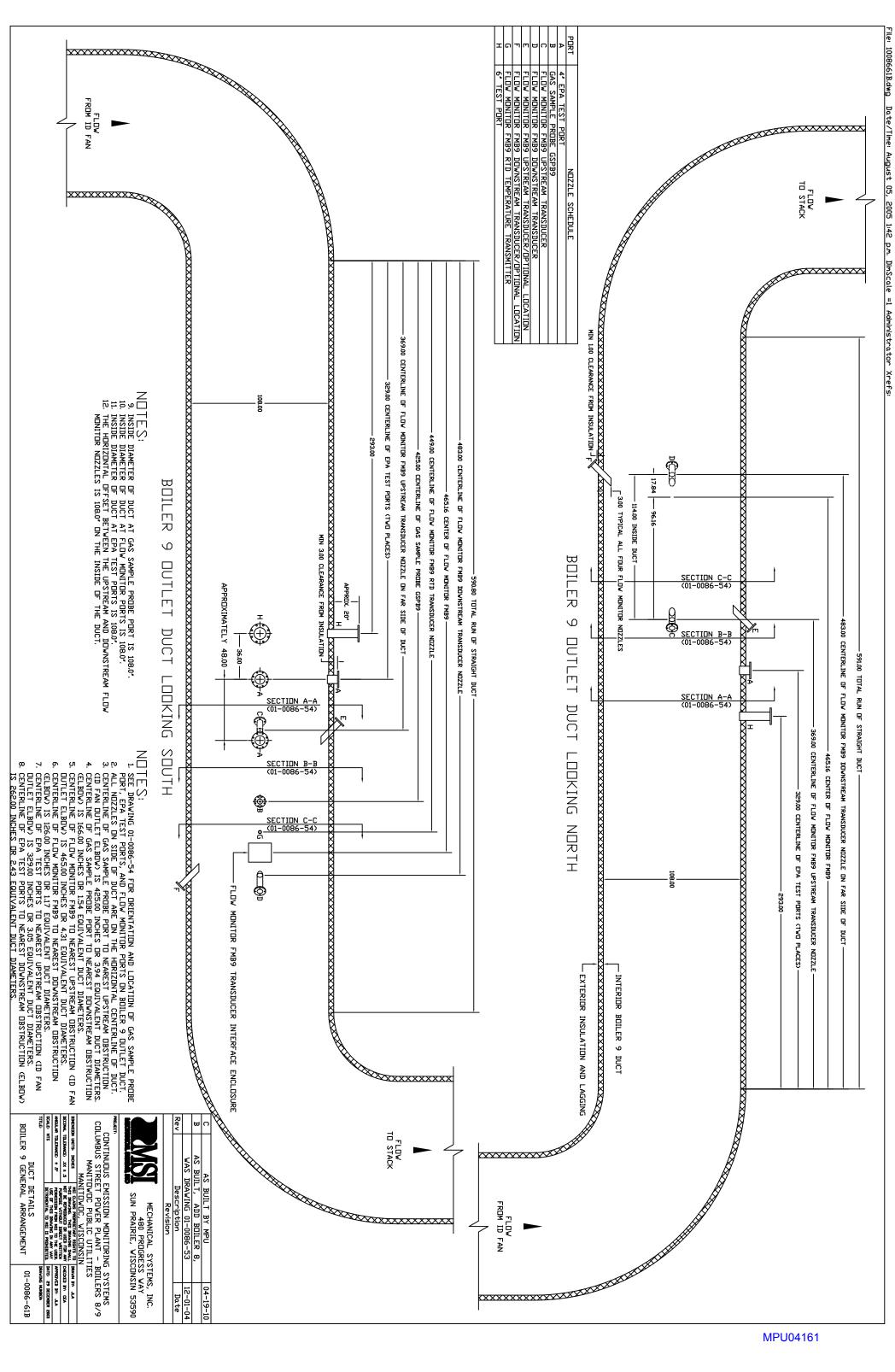
Boiler 9 (B09) has a 9-foot diameter exhaust. The diagram for the unit's ductwork is attached with the B09 method 1 in the appendix. The drawings include the testing platform and test port details.

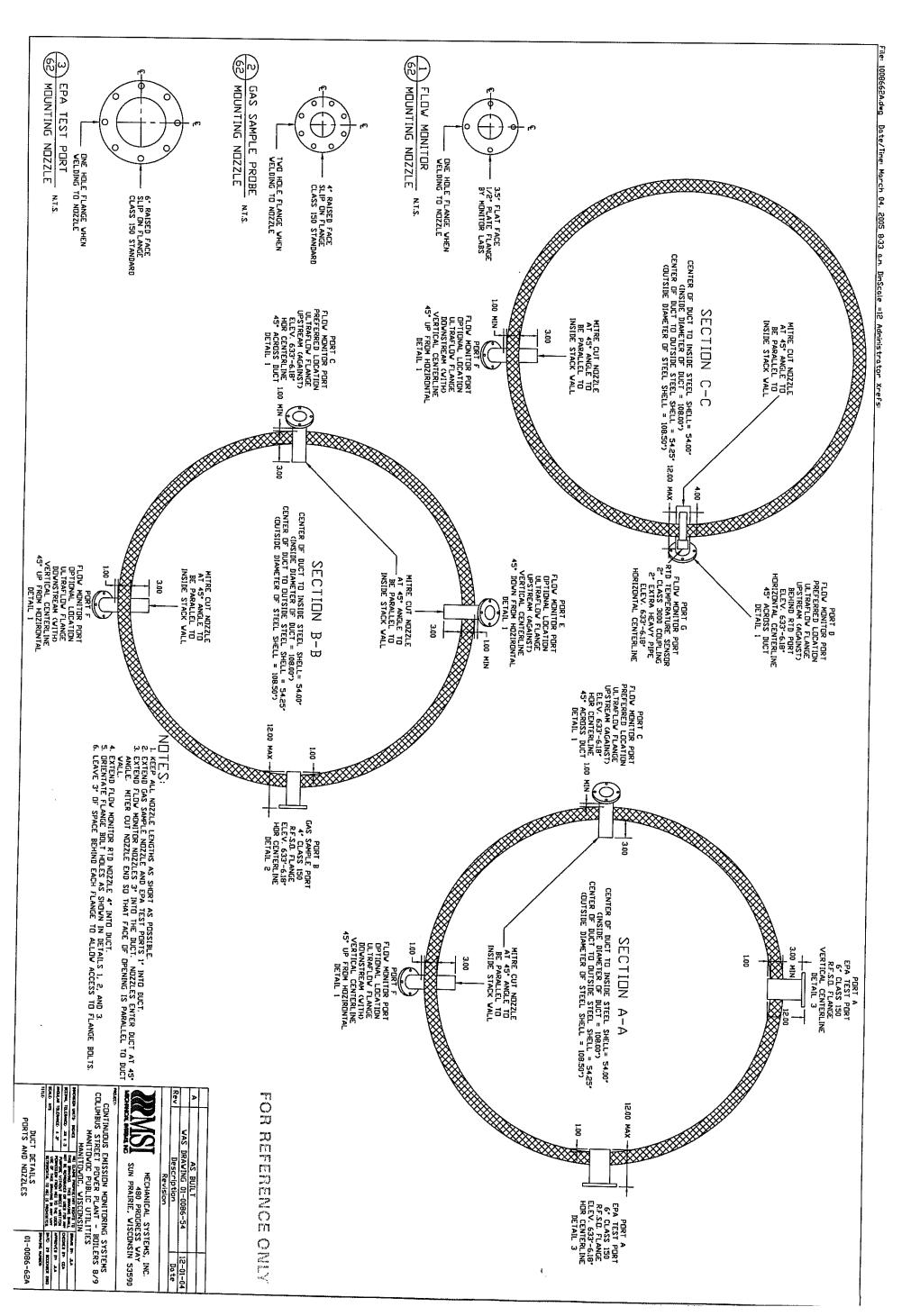
# **Appendix**

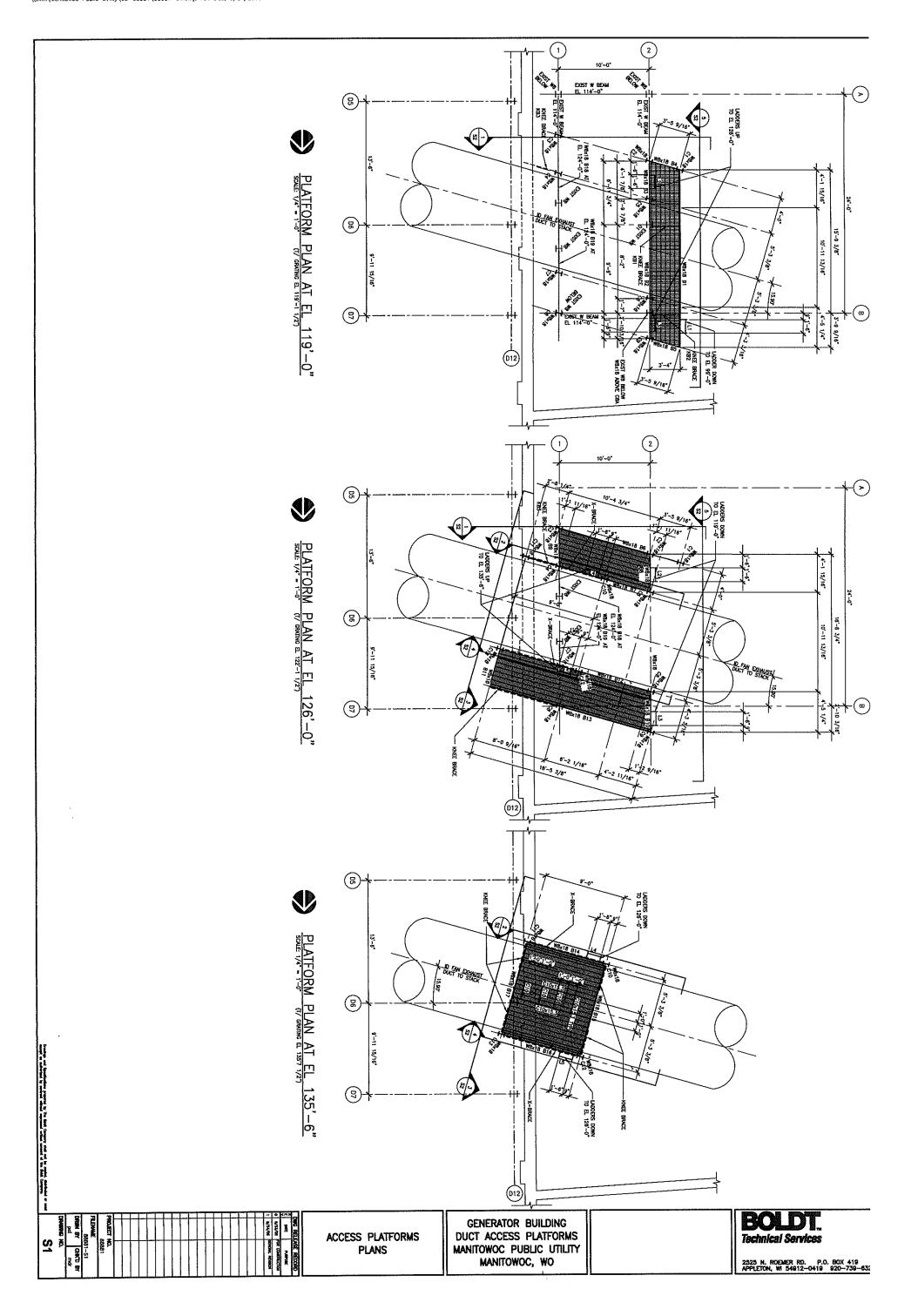
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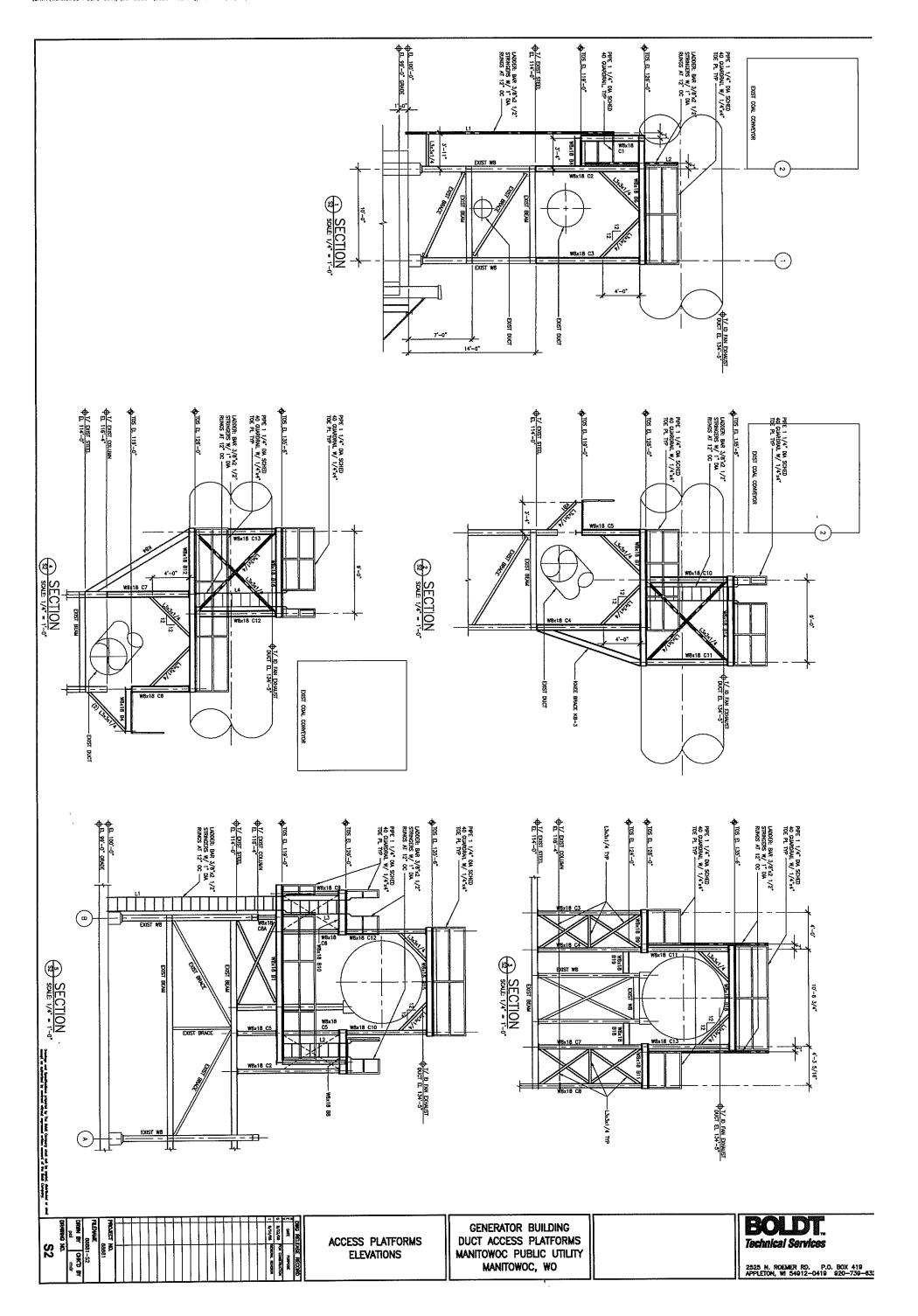


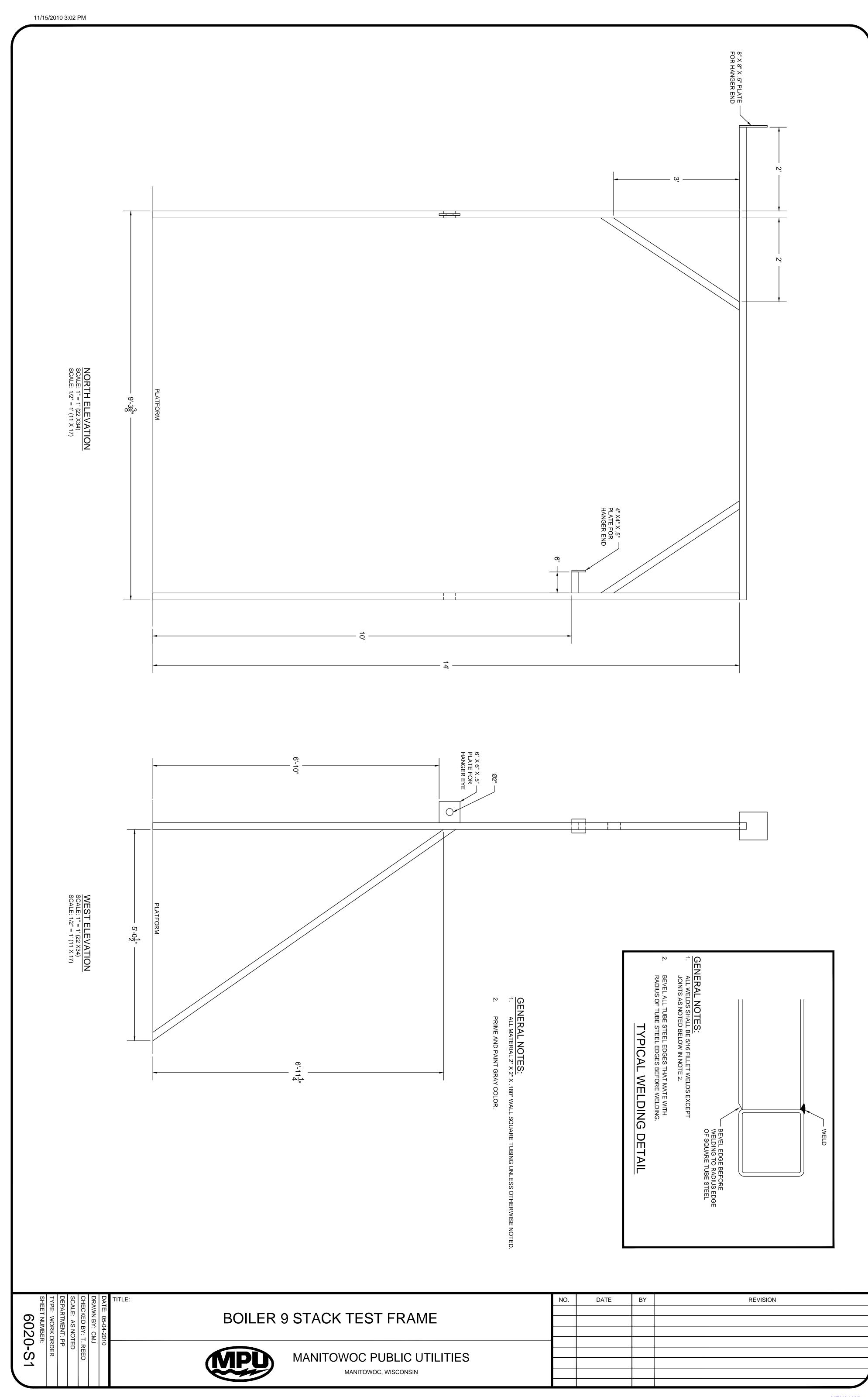


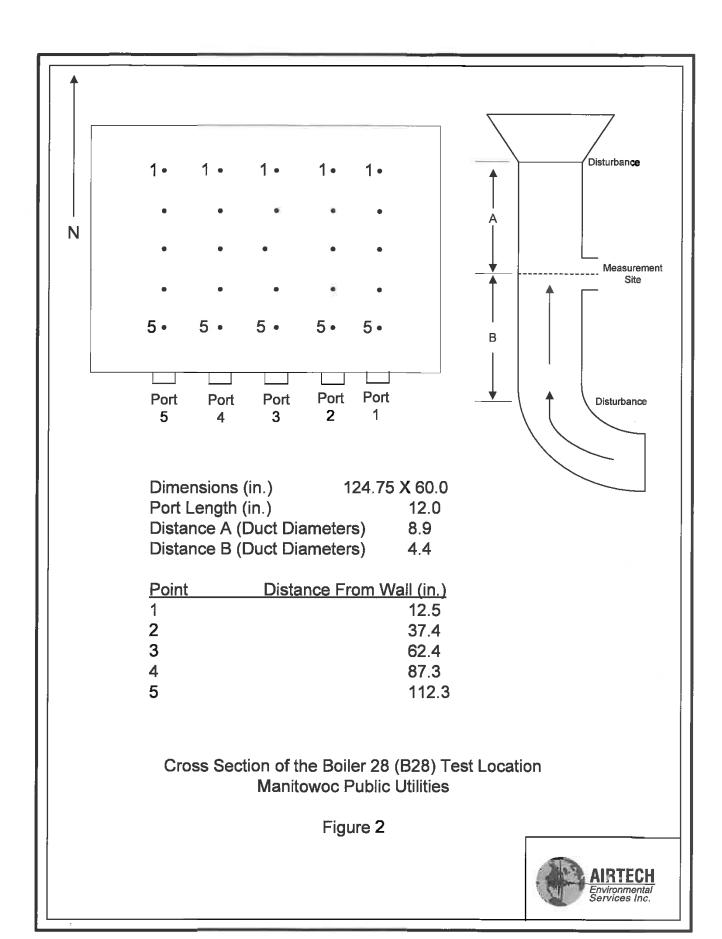


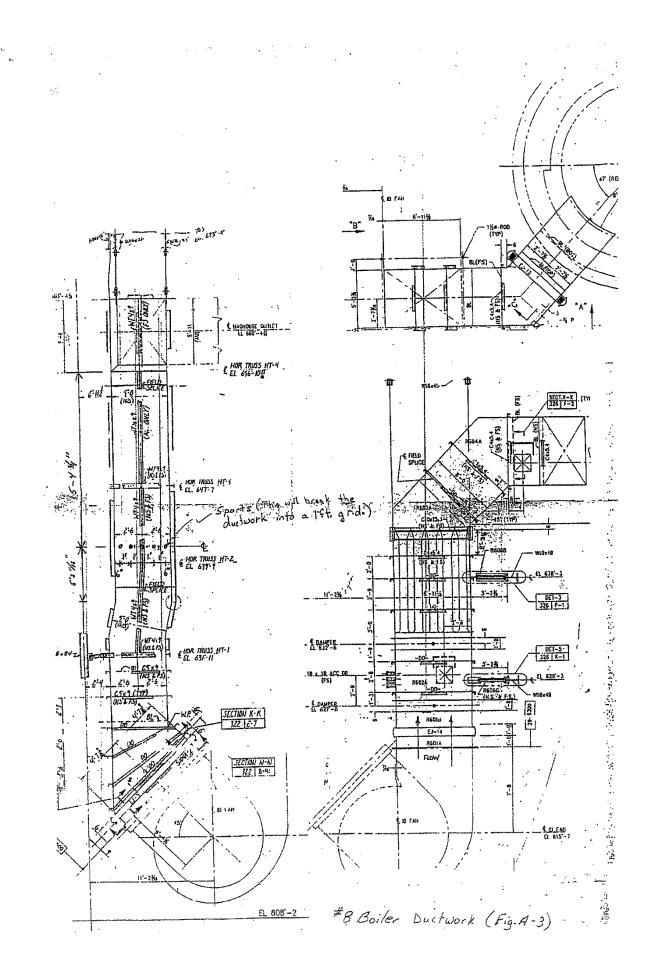


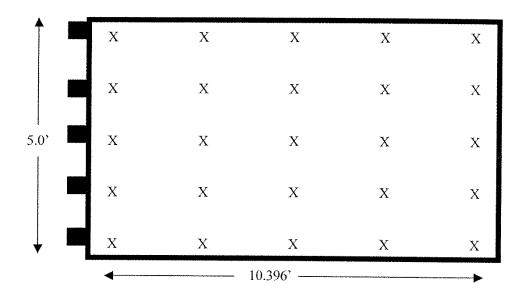












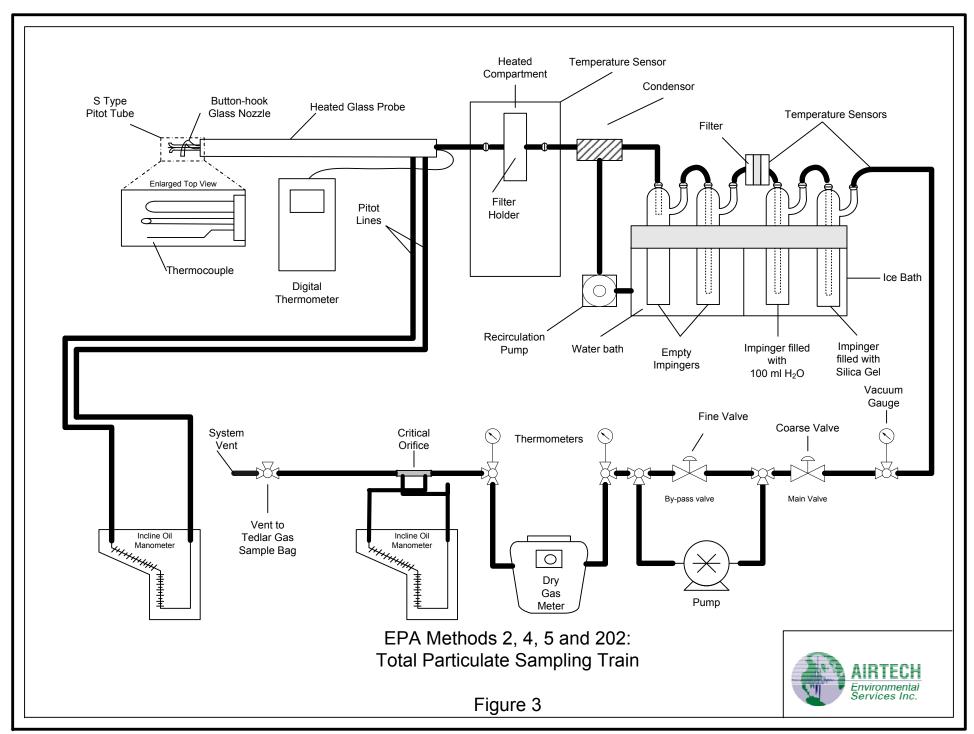
Job: Manitowoc Public Utilities Manitowoc, Wisconsin

Area: 51.979 Square Feet

Unit Location: Boilers B28 No. Test Ports: 5

Length: 10 Feet 4.75 Inches Tests Points per Port: 5

Width: 5.0 Feet



Sample Calculations

### **Area of Sample Location**

$$\mathbf{A}_{\mathrm{s}} = \pi \times \left(\frac{\mathrm{d}}{2}\right)^2$$

where:

 $A_s$  = area of sample location (ft<sup>2</sup>)

d = stack diameter (ft)

2 = conversion factor (diameter to radius)

### **Stack Pressure Absolute**

$$P_{a} = P_{b} + \frac{P_{s}}{13.6}$$

where:

P<sub>a</sub> = stack pressure absolute (in. Hg) P<sub>b</sub> = barometric pressure (in. Hg) P<sub>s</sub> = static pressure (in. H<sub>2</sub>O)

13.6 = conversion factor (in.  $H_2O/in Hg$ )

### Volume of Dry Gas Collected Corrected to Standard Temperature and Pressure

$$V_{m(std)} = \frac{17.64(V_{m})(Y_{d})(P_{b} + \frac{\Delta H}{13.6})}{(T_{m} + 460)}$$

where:

 $V_{m(std)}$  = volume of gas collected at standard temperature and pressure (scf)

V<sub>m</sub> = volume of gas sampled at meter conditions (ft<sup>3</sup>) Y<sub>d</sub> = gas meter correction factor (dimensionless)

P<sub>b</sub> = barometric pressure (in. Hg)

 $\Delta H$  = average sample pressure (in. H<sub>2</sub>O)  $T_m$  = average gas meter temperature (°F) 13.6 = conversion factor (in. H<sub>2</sub>O/in Hg)

17.64 = ratio of standard temperature over standard pressure (°R/in.Hg)

= conversion ( $^{\circ}$ F to  $^{\circ}$ R)

# Volume of Water Vapor Collected Corrected to Standard Temperature and Pressure

$$V_{w(std)} = 0.04707 \times V_{wc} + 0.04715 \times V_{wsg}$$

where:

 $V_{w(std)}$  = volume of water vapor at standard conditions (scf)

 $V_{wc}$  = weight of liquid collected (g)  $V_{wsg}$  = weight gain of silica gel (g)

0.04707 = volume occupied by one milliliter of water at standard temperature and

pressure (ft<sup>3</sup>/ml)

0.04715 = volume occupied by one gram of water at standard temperature and

pressure  $(ft^3/g)$ 

### Percent Moisture<sup>1</sup>

$$\mathbf{B}_{\mathrm{ws}} = 100 \times \left[ \frac{\mathbf{V}_{\mathrm{w(std)}}}{\left( \mathbf{V}_{\mathrm{m(std)}} \right) + \mathbf{V}_{\mathrm{w(std)}}} \right]$$

where:

 $B_{ws}$  = moisture content of the gas stream (%)

 $V_{m(std)}$  = volume of gas collected at standard temperature and pressure (scf)

 $V_{w(std)}$  = volume of water vapor at standard conditions (scf)

100 = conversion factor

# Molecular Weight of Dry Gas Stream<sup>2</sup>

$$\mathbf{M}_{d} = \left(44 \times \frac{\% \text{CO}_{2}}{100}\right) + \left(32 \times \frac{\% \text{O}_{2}}{100}\right) + \left(28 \times \frac{(\% \text{N}_{2})}{100}\right)$$

where:

 $M_d$  = molecular weight of the dry gas stream (lb/lb-mole)

%CO<sub>2</sub> = carbon dioxide content of the dry gas stream (%)

= molecular weight of carbon dioxide (lb/lb-mole) %O<sub>2</sub> = oxygen content of the dry gas stream (%)

= molecular weight of oxygen (lb/lb-mole) %N<sub>2</sub> = nitrogen content of the dry gas stream (%)

28 = molecular weight of nitrogen (lb/lb-mole)

100 = conversion factor

<sup>&</sup>lt;sup>1</sup> The moisture saturation point will be used for all calculations if it is exceeded by the actual moisture content.

<sup>&</sup>lt;sup>2</sup> The remainder of the gas stream after subtracting carbon dioxide and oxygen is assumed to be nitrogen.

### Molecular Weight of Wet Gas Stream

$$\mathbf{M}_{s} = \left(\mathbf{M}_{d} \times \left(1 - \frac{\mathbf{B}_{ws}}{100}\right)\right) + \left(18 \times \frac{\mathbf{B}_{ws}}{100}\right)$$

where:

M<sub>s</sub> = molecular weight of the wet gas stream (lb/lb-mole) M<sub>d</sub> = molecular weight of the dry gas stream (lb/lb-mole)

B<sub>ws</sub> = moisture content of the gas stream (%) 18 = molecular weight of water (lb/lb-mole)

100 = conversion factor

### **Velocity of Gas Stream**

$$V_{s} = 85.49 \left(C_{p}\right) \left(\sqrt{\overline{\Delta P}}\right) \sqrt{\frac{\left(T_{s} + 460\right)}{\left(M_{s}\right)\left(P_{b} + \frac{P_{s}}{13.6}\right)}}$$

where:

V<sub>s</sub> = average velocity of the gas stream (ft/sec) C<sub>p</sub> = pitot tube coefficient (dimensionless)

 $\sqrt{\Delta P}$  = average square root of velocity pressures (in. H<sub>2</sub>O)<sup>1/2</sup>

Ts = average stack temperature (°F)

Ms = molecular weight of the wet gas stream (lb/lb-mole)

Pb = barometric pressure (in. Hg)

Ps = static pressure of gas stream (in.  $H_2O$ )

85.49 = pitot tube constant (ft/sec)([(lb/lb-mole)(in. Hg)]/[( ${}^{\circ}$ R)(in. H<sub>2</sub>O)])  ${}^{1/2}$ 

460 = conversion (°F to °R)

13.6 = conversion factor (in.  $H_2O/in Hg$ )

### **Volumetric Flow of Gas Stream - Actual Conditions**

$$Q_a = 60(V_s)(A_s)$$

where:

Q<sub>a</sub> = volumetric flow rate of the gas stream at actual conditions (acfm)

 $V_s$  = average velocity of the gas stream (ft/sec)

A<sub>s</sub> = area of duct or stack (ft<sup>2</sup>) 60 = conversion factor (sec/min)

### **Volumetric Flow of Gas Stream - Standard Conditions**

$$Q_{std} = \frac{17.64(Q_a)\left(P_b + \frac{P_s}{13.6}\right)}{(T_s + 460)}$$

where:

 $Q_{std}$  = volumetric flow rate of the gas stream at standard conditions (scfm) = volumetric flow rate of the gas stream at actual conditions (acfm)

T<sub>s</sub> = average stack temperature (°F) P<sub>b</sub> = barometric pressure (in. Hg)

P<sub>s</sub> = static pressure of gas stream (in. H<sub>2</sub>O) 13.6 = conversion factor (in. H<sub>2</sub>O/in Hg)

17.64 = ratio of standard temperature over standard pressure (°R/in.Hg)

= conversion ( $^{\circ}$ F to  $^{\circ}$ R)

### Volumetric Flow of Gas Stream - Standard Conditions - Dry Basis

$$Q_{dstd} = Q_{std} \left( 1 - \frac{B_{ws}}{100} \right)$$

where:

 $Q_{dstd}$  = volumetric flow rate of the gas stream at standard conditions, on a dry

basis (dscfm)

 $Q_{std}$  = volumetric flow rate of the gas stream at standard conditions (scfm)

 $B_{ws}$  = moisture content of the gas stream (%)

100 = conversion factor

### Area of Nozzle

$$A_{n} = \pi \times \left(\frac{d_{n}}{2 \times 2}\right)^{2}$$

where:

An = area of nozzle (ft2) dn = diameter of nozzle (in) 12 = conversion factor (in/ft)

2 = conversion factor (diameter to radius)

### **Percent Isokinetic**

$$I = \frac{0.0945 (T_s + 460) (V_{m(std)})}{\left(P_b + \frac{P_s}{13.6}\right) (V_s) (A_n) (\Theta) \left(1 - \frac{B_{ws}}{100}\right)}$$

where:

I = percent isokinetic (%)

 $T_s$  = average stack temperature (°F)

 $V_{m(std)}$  = volume of gas collected at standard temperature and pressure (scf)

P<sub>b</sub> = barometric pressure (in. Hg)

P<sub>s</sub> = static pressure of gas stream (in. H<sub>2</sub>O) V<sub>s</sub> = average velocity of the gas stream (ft/sec)

 $A_n$  = cross sectional area of nozzle (ft<sup>2</sup>)

 $\Theta$  = sample time (min)

 $B_{ws}$  = moisture content of the gas stream (%)

0.0945 = conversion ( ${}^{\circ}F$  to  ${}^{\circ}R$ )

13.6 = conversion factor (in.  $H_2O/in Hg$ )

100 = conversion factor

### **Acetone Wash Blank**

$$W_{a} = \frac{\left(m_{ab}\right)\left(v_{aw}\right)}{v_{ab}}$$

where:

W<sub>a</sub> = particulate mass in acetone wash blank (g) m<sub>ab</sub> = mass collected, acetone wash blank (g)

 $v_{aw}$  = volume of wash (ml)

 $v_{awb}$  = volume of acetone wash blank (ml)

### **Total Particulate Catch**

$$M_n = m_f + (m_a - W_a)$$

where:

 $M_n$  = total particulate catch (g)  $m_f$  = particulate on filter (g)  $m_a$  = particulate in wash (g)

 $W_a$  = particulate mass in acetone wash blank (g)

### Particulate Concentration, grains/dscf

$$C = \frac{\left(M_n\right)\left(5 . 3\right)}{V_{m \text{ sd}}}$$

where:

C = particulate concentration (grains/dscf)

 $M_n$  = total particulate catch (g)

 $V_{m(std)}$  = volume of gas collected at standard temperature and pressure (dscf)

15.43 = conversion factor (grains/gram)

### Particulate Emission Rate (lb/hr)

$$E_{\text{thr}} = \frac{(M_n)(Q_{\text{old}})(60)}{(V_{\text{mod}})(456)}$$

where:

 $E_{lb/hr}$  = particulate emission rate (lb/hr)

 $M_n$  = total particulate catch (g)

 $V_{m(std)}$  = volume of gas collected at standard temperature and pressure (scf)  $Q_{dstd}$  = volumetric flow rate of the gas stream at standard conditions, on a dry

basis (dscfm)

60 = conversion factor (min/hr) 453.6 = conversion factor (g/lb) Sample Calibration Sheets

# Airtech Environmental Services, Inc.

Meter Box Full Test Calibration

Date: Operator:

Meter Bo	x ID			Meter Box	¢ΔH@	H@ Meter Box Y <sub>d</sub> Barometric				ric Pressure (in. Hg.)					
Time		Orifice	Data			Meter Box Data				Results					
θ (min)	K'	Vacuum	$T_{amb}$	$V_{cr}$	$V_{initial}$	$V_{final}$	$V_d$	ΔΗ	$T_i$	$T_o$	T <sub>avg</sub>	$V_{mstd}$	Q	$Y_d$	∆H@
													Average		

	Nomenclature						
K'	Critical Orifice Coefficient						
$T_{amb}$	Ambient Temperature (°F)						
$V_{cr}$	Volume Through Orifice (scf)						
$V_{d}$	Gas Meter Volume (ft <sup>3</sup> )						
ΔΗ	Orifice Pressure Differential (in. H <sub>2</sub> O)						
T <sub>i</sub>	Meter Inlet Temperature (°F)						
T <sub>o</sub>	Meter Outlet Temperauture (°F)						
$T_{avg}$	Average Meter Box Temperature (°F)						
$V_{mstd}$	Volume Metered Standardized (scf)						
Q	Flow Rate (scfm)						
Y <sub>d</sub>	Meter Correction Factor (dimensionless)						
ΔH@	ΔH yielding 0.75 scfm						

Vacuum Guage (in. Hg.)		Thermometers (°F)			Equations
Standard	Vacuum	Standard	DGM	DGM	
	Gauge		Inlet	Outlet	$V_{cr} = K' * P_b * \theta$
					$V_{cr} = \frac{K' * P_b * \theta}{(T_{amb} + 460)} ^ 0.5$
					V = 17.64 *V * (D + (AU/12.6))
					$V_{\text{mstd}} = \frac{17.64 * V_{\underline{d}} * (P_{\underline{b}} + (\Delta H/13.6))}{(T_{\text{avg}} + 460)}$
					$Q = V_{cr} / \theta$
					$Y_d = V_{cr} / V_{mstd}$
					$\Delta H @ = \underline{.0319 * \Delta H * (T_{avg} + 460) * \theta^{2}}{P_{b} * Y_{d}^{2} * V_{m}^{2}}$
	·				- <b>u</b>

# **Airtech Environmental**

# **Post Test Meter Calibration**

Average Field Sample Rate (cfm)	Date	
Highest Field Vacuum (inches Hg)	Client	
Critical Orifice ID	Project No.	
Orifice Flow Rate (cfm)	Meter ID	

	Run 1	Run 2	Run 3	
Initial Volume (ft <sup>3</sup> )				
Final Volume (ft <sup>3</sup> )				
Volume Metered (ft <sup>3</sup> )				
DGM Inlet Temperature (°F)				
DGM Outlet Temperature (°F)				
Average DGM Temperature (°F)				
Ambient Temperature (°F)				
Elapsed Time (min.)				
ΔH (inches H <sub>2</sub> O)				
Barometric Pressure (inches Hg)				
Pump Vacuum (inches Hg)				
K'				
Vcr (ft <sup>3</sup> )				
Vmstd (ft <sup>3</sup> )				
Post Test Yc				
Full Test Yd				
% Difference				
	Average Differ	ence		

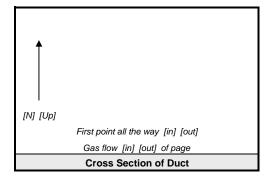
Sample Data Sheets

# Airtech Environmental Services, Inc.

Method 1 Data Sheet

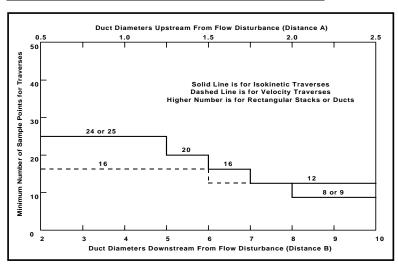
LOCATION
----------

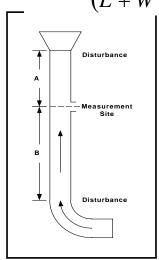
Client		
Project No:		
Plant		
Date		
Technician		
<b>Duct Diameter</b>	(in.)	
Port Diameter (	in.)	
Port Length (in	.)	
Port Type		
Distance A (ft)		
Distance B (ft)		
Distance A (Duct Diameters)		
Distance B (Du	ct Diameters)	



For rectangular ducts

$$ED = \frac{2LW}{(L+W)}$$





	Traverse	Distance
Location Schematic and Notes	Point	(in.)
	1	
	2	
	3	
	4	
	5	
	6	
	7	
	8	
	9	
	10	
	11	
	12	
	13	_
	14	
Indicate sample ports, height from grade, types of disturbances, access, unistrut configuration, etc.	15	
Distance to point must include length of port	16	

### AIRTECH ENVIRONMENTAL SERVICES INC.

Oxygen and Carbon Dioxide Data Sheet

PROJECT	NO.	i		Page		of		
				•				
Client								
Plant					$F_0 = (20.9)$	·O <sub>2</sub> %)		
_ocation		Date			CC	2%		
Analyzer Type		Leak Check						

Run No.	Trial No.	%CO <sub>2</sub>	%CO <sub>2</sub> +%O <sub>2</sub>	%O <sub>2</sub>	Fo	Analyst	Date	Time
	1							
	2							
	3							
	Average				<u></u>			
	1							
	2					-	-	-
	3				7			
	Average				<u> </u>			
	1							
	2							
	3							
	Average							
	1							
	2							
	3							
	Average							
	1							
	2							
	3							
	Average							
	1							
	2							
	3							
	Average							
	1							
	2							
	3							
	Average							
	1							
	2							
	3							
	Average							
	1							
	2					-	-	-
	3				7			
	Average				<u></u>			
	1							
	2					-	-	-
	3		_					
	Average							

Notes:

Measurements must be made to the nearest 0.2%.

Three different trails should be performed for each sample.

The differences between the trials must not be greater than 0.2% overall.

Run an ambient air check to verify Oxsorb.

Expected F<sub>o</sub> Ranges

Anthracite/Lignite 1.015-1.130 Nat. Gas 1.600-1.836 Bituminous 1.083-1.230 Wood Bark 1.000-1.120

Distillate Oil 1.260-1.413 Municiple

Residual Oil 1.210-1.370 Garbage 1.043-1.177

MPU04182

## AIRTECH ENVIRONMENTAL SERVICES INC.

General Testing Data Sheet

						TE	ESTING T	YPE:				_			
RUN	NO			_			MET	HOD NO			_			Page	of
Client												Barometr	ic (inHg)	Water [ml] [g]	
Plant							1					Ambient '	Temp (⁰F)	Silica gel (g)	
Location			<b>1</b> ↑					Static (inH <sub>2</sub> O)		Total VIc					
Date Project No.							Probe ID		Liner Type						
Meter Ope	Meter Operator							Nozzle ID		Nozzle Dia (in)					
Probe Operator			Filter ID												
Meter ID		Yd			Pitot Cp		[N] [Up]					Train ID		Train Type	
ΔH@	Kf Leak check				First point all the way [in] [out] Duct Dim. (in)						Port Lngth (in)				
Pre Leak	Check		[cfm] [lpn	n] @		(inHg)		Gas fle	ow [in] [out]	of page					
Post Leak Check [cfm] [lpm] @ (inHg)		Cross Section of Duct					Start Time		Stop Time						
	Min/Point	Velocity	Orifice	Ga	s Sample		Probe	Filter	Impinger	DGM	DGM				
		Pressure			/olume	Stack	Temp	Temp	Outlet	Inlet	Outlet	Pump	Auxiliary		
Traverse	Elapsed	ΔΡ	ΔН	Initi	ial [f³] [l]	Temp	(°F)	(ºF)	Temp	Temp	Temp	Vacuum	Temp		
Point	Time	(inH <sub>2</sub> O)	(inH <sub>2</sub> O)			(°F)			(°F)	(ºF)	(°F)	(inHg)	(°F)	Notes	
Total															
Average															